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## **Chapter 5 Review The Periodic**

CHAPTER 5 REVIEW The Periodic Law  
SECTION 3 SHORT ANSWER Answer the following questions in the space provided.

1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released.

2. d The energy required to remove an electron from a neutral atom is the atom's

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SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design.

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SECTION 5-3. SHORT ANSWER Answer the following questions in the space provided. 1. When an electron is added to a neutral atom, energy is . (a) always absorbed (c) either absorbed or released. (b) always released (d) burned away. 2.

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SHORT ANSWER Answer the following questions in the space provided. 1. Consider the neutral atom with 53 protons and 74 neutrons to answer the following questions. 53. a. What is its atomic number? 127. b. What is its mass number? atomic number. c. Is the element's position in a modern periodic table ...

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(36) The order of elements in the periodic table is based on atomic mass the number of protons in the nucleus the number of neutrons in the nucleus

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Science Chapter 5 The Periodic Table. The atomic mass of an element is. Which statement is true about the metal.... At room temperature none of the metals.... Which general statement does NOT apply.... a weighted average of the masses of an element's isotopes. Silicon is a better conductor of electric current than sulfur....

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The energy required to remove an electron from an atom is the atom's .

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## **Chapter 5 Review The Periodic Law Answers 5 1**

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## Answer Key

Answers 5 1 the modern periodic law states that the physical and chemical properties of an element are functions of its atomic number the discovery of the noble gases changed

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SECTION 5-3 SHORT ANSWER Answer the following questions in the space provided 1 When an electron is added to a neutral atom, energy is (a) always absorbed (c) either absorbed or released (b) always released (d) burned away 2 The energy required to remove an electron from an atom is the

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(b) according to Mendeleev's original design.

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