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secondary students. Avogadro's number is one of the culprits. Although it is possible to do a slew of factor label problems with Avogadro's number, it is not the primary value of the mole concept in chemistry. This.

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It's jumbled up in mixtures that Most chemicals abound.

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Answer the questions below based on your reading above and on your knowledge of chemistry. Answer questions 1-4 by referring to the diagram to the right showing an electrochemical cell. The metal at electrode A is silver. The metal at electrode C is lead. The electrolytes at locations B, D, and E are potassium nitrate, silver nitrate,

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Sample Problem 1 What is the concentration of a 30. mL sample of HCl if it can be neutralized by 50. mL of 1.2 M NaOH? Step 1: Note the ratio of H<sup>+</sup> to OH<sup>-</sup> is 1 to 1 Step 2: Substitute values into

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the equation  $M_A \times V_A = M_B \times V_B$   $M_A(30. \text{ mL}) = (1.2 \text{ M})(50. \text{ mL})$  Step 3: Solve for the unknown M  
12M 50mL

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