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thermochemistry. the study of the transfers of energy as heat that accompany chemical reactions and physical changes. calorimeter. a device used to measure the energy absorbed or released as heat in a chemical or physical change.

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ANSWER Answer the following questions in the space provided. 1.

Describe Hess's law.

The overall enthalpy change in a reaction is equal to the sum of enthalpy changes for the individual steps in the process. 2. What

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determines the amount of energy absorbed by a material when it is heated?

16 Reaction Energy - David Brearley High School

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ANSWER Write the answers to the following questions in the space provided. 22.

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