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OF
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 2. SOME
DEFINITIONS
A solution is a _____

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mixture of 2 or more substances in a single phase.
One constituent is usually regarded as the SOLVENT and the others as SOLUTES.
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Chemistry Notes 2. Solutions • Basic definitions – Solution: homogenous mixture – Solute: substance that is dissolved – Solvent: substance that contains the solute in a solution • Process of Dissolving: – Molecules of the solvent “pull apart” the solute molecules, and distributes them within the solvent • How to tell the solute from

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 $Zn + 2H^+ \rightarrow Zn^{2+} + H_2$ $NH_3 + H_2O \rightarrow NH_4^+$

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+ OH- HClO + LiOH LiClO + H₂O HCl +
H₂O H₃O+ + Cl- pH Scale ranges from
0-14 Acids found between a number
close to 0 & 7 Bases found between 7 &
14 Neutral 7 measures the acidity or
basicity of a solution by focusing on the
concentration of hydrogen ions (H⁺) in a
solution equals the negative log of the
concentration of H⁺ Mixtures (Solutions)

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Mixtures a combination of two or more substances that do not combine chemically, but remain the same individual ...

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mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

Chemistry Notes for class 12

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Chapter 2 Solutions

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An unsaturated solution is a solution in which a solvent is capable of dissolving any more solute at a given temperature. A saturated solution can be defined as a solution in which a solvent is not capable of dissolving any more solute at a given temperature. The solutions are of two

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forms, depending on whether the solvent is water or not.

Types of Solutions - Different Types, Homogeneous ...

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Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more ...

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This slideshow explains solution vocabulary (solute, solvent, suspension, solution, miscible, etc.), the process of

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dissolving with heats of solution, mole fractions, molarity, and gases dissolved in water. There are practice problems embedded in each powerpoint for student interaction, which include helpful tips for each problem. Related Products:

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